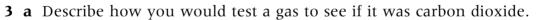


## 7Fc/4 Reactions of acids 2

Acids can be corrosive. Acids react with many substances to form new materials. One of these materials is often a gas.



- 1 a How would you know if a gas was being produced in a reaction?
  - **b** Copy this test tube, and draw what you would see if the reaction was producing a gas.
- **2 a** Describe how you would test a gas to see if it was hydrogen.
  - **b** How would you know whether or not the gas was hydrogen?
  - **c** Write a list of the apparatus you would need.
  - **d** What safety precautions should you take when doing this test?



- **b** How would you know whether or not the gas was carbon dioxide?
- **4** Look at the list of substances below.
  - **a** Write down two substances that would react with an acid to produce hydrogen gas.
  - **b** Write down two substances that would react with an acid to produce carbon dioxide gas.

calcium oxide calcium carbonate iron oxide

magnesium magnesium carbonate magnesium oxide

aluminium sodium chloride

- **5** Rainwater is a weak acid.
  - **a** What can rainwater do to metals like iron?
  - **b** What can rainwater do to rocks like limestone?

